

**General information**

**File No.:** 40

**Technology:** Electrooxidation

**Typology:** Liquid fraction processing



**Goal:** Oxidation of organic components, metals, etc., in the anode of an electrochemical reactor through the application of an external electric current

**TRL:** 4

**Status:** Consolidated

**Complexity:** High

**Inlets:**

Water flow from other processes (separation by pressing -> electrocoagulation -> Electrooxidation)

**Products:**

Treated water

**Other outputs:**

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**Consumption**

**Energy:** High

**Water:** Not applicable

**Reagents:** High

**Efficiency:**

Electrooxidation combined with electrocoagulation: Reduction of total nitrogen by 99.9%; Reduction of ammoniacal nitrogen by 99.9%; 99.3% COD reduction

**Economics**

**Investment:**

Not available

**Operation:**

Wastewater from mills: €0.22-1.12/kg DQO Minimal  
Wastewater SL: €1.9/m<sup>3</sup>(pork porridge)

**Strong points:**

High removal of COD and conversion of nitrogen to ammoniacal nitrogen;  
High reduction of recalcitrant organic compounds (phenols).

**Weak points:**

Potential risk of ammonia emissions due to conversion to organic nitrogen;  
Possible formation of chlorinated organic compounds in those indirect oxidation processes, and contamination by added metals;  
Colloidal organic matter can produce interference.

**Others:**

The material used for the anode is important. Mainly, the most typical materials include glassy carbon, Tu/RuO<sub>2</sub>, Ti/Pt-Ir, carbon fiber, MnO<sub>2</sub>, Pt-carbon black, porous carbon with stainless steel filter and reticulated vitreous carbon

**Technology train:**

1. Separation by pressing -> Electrocoagulation -> Electrooxidation.